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Section 7.3 – Bonding in Metals.

The valence electrons of metal  
atoms can be modeled as a sea of  
electrons. Metallic bonds consist

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of the attraction of the free-floating valence electrons for the positively charged metal ions. Metals are good conductors and malleable because of their mobile electrons.

Chapter 7 – Ionic and Metallic

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## Bonding Answer Key

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1. Chapter 7 "Ionic and Metallic Bonding" 2. Section 7.1 - Ions  
OBJECTIVES: -Determine the number of valence electrons in an atom of a representative element.  
3. Section 7.1 - Ions OBJECTIVES:  
-Explain how the octet rule applies to atoms of metallic and

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nonmetallic elements. 4.

Chemistry - Chp 7 - Ionic and  
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206 Chapter 7 □ Ionic Compounds  
and Metals Section 77.1.1 Figure  
7.1 As carbon dioxide dis-solves  
in ocean water, carbonate ions

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are produced. Coral polyps capture these car-bonate ions, producing crystals of calcium carbonate, which they secrete as an exo-skeleton. Over time, the coral reef forms. A coral reef is a complex habitat that

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Chapter 7: Ionic Compounds and  
Metals

Chapter 7 Ionic and Metallic  
Bonding Multiple Choice Identify  
the choice that best completes  
the statement or answers the  
question. \_\_\_\_ 1. What are the  
valence electrons of an atom? a.

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**Answer Key**  
The electrons that an atom gains when it becomes an anion b. The electrons that remain closest to the nucleus c. The electrons in the highest occupied energy level d. The electrons that an atom loses when it ...

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an element's.... diagrams that

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show valence electrons in the  
atoms of an elemen.... in forming  
compounds, atoms tend to  
achieve the electron confi....

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Lesson 7.1 Reading and Study

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Essential Understanding Ions form when atoms gain or lose valence electrons, becoming electrically

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Answer Key  
charged. Lesson Summary.

Valence Electrons Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation. For a representative element, the group number equals the ...

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## BONDING AND INTERACTIONS

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L. Cotton. Section 7.1 - Ions

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number of valence electrons in an  
atom of a representative element.

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charge Chapter 8 Energetics of  
bond formation and dissociation  
Bond formation:  $A + B \rightarrow AB$  If the



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reaction is energetically favorable, AB is lower in energy than A + B. Heat is released .

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