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the space provided, write a definition for each of the terms 3.

Chemical reactions always involve changes in chemical. Energy in Reactions (page 50) 4. What is released or absorbed whenever chemical bonds form or are broken? 5. What do chemical reactions that absorb energy need ...

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(water) = $100 \text{ ml} * 1.03 \text{ g/ml} = 103 \text{ g}$ $q(\text{reaction}) = -m * T * C(\text{water})$
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