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3—3 Review and Reinforcement Modern Atomic Theory Complete the following table. 1. 2. 3. If the statement is true, write “true.” If it is false, change the underlined word or words to make the statement true. Write your answer on the line provided. 4. Moseley discovered that all atoms of an element have the same number of neutrons in ...

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Check the following equalities for errors. If an answer is correct, write "correct" in the space provided. If an answer is incorrect, rewrite it to make it correct. 21.  $= 4.563 \times 10^7$  22.  $0.000\ 253 = 2.53 \times 10^{-3}$  23.  $= 68.05 \times 10^8$  Solve each of the following problems as directed. Show all your work. 24. 25. 26. 34 An unknown liquid has a mass of ...

[Quia](#)

2 – 3 Review and Reinforcement -Matter-Type your answers in Red Complete the following sentences by filling in the appropriate word from the list below. Gas Liquid Solid Plasma Matter Energy Physical Chemical Evaporation Matter is anything that has mass and volume. 1. The two states of matter that occupy a definite volume are solid and liquid. gas and plasma are the other two states of matter.

[4. 2\u0020133 Review and Reinforcement - Matter - 4 23 ...](#)

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7-2 Review and Reinforcement. Covalent Bonding. If the statement is true, write “true.” If it is false, change the underlined word or words to make it true. Write your answer on the line provided. \_\_\_\_F, covalent\_ 1. A group of atoms united by ionic bonds is called a molecule. \_\_\_\_T\_\_\_\_ 2. A covalent bond is formed by a shared pair of electrons

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7—2 Review and Reinforcement Covalent Bonding C ass If the statement is true, write "true." If it is false, change the underlined word r words to make it true. Write your answer on the line provided. 1. 2. 3. 4. 5. 8. A group of atoms united by ionic bonds is c led a molecule. A covalent bond is formed by a shared pair f electrons.

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6—2 Review and Reinforcement (continued) Identify each of the following elements as a meta/ (M), nonmetal (NM), or semimetal (SM). 19. sodium —SM— silicon 21. neon 22. 23. nitrogen Write the family names that have been given to each of the following groups. 24. Group IA alkan- 25. Group 2A 26. Group 7 A 27. Group8A hvohlt

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